## ALDOL REACTION OF ALUMINIUM ENOLATE RESULTING FROM 1,4-ADDITION OF Me<sub>2</sub>AlSPh TO $\alpha,\beta$ -UNSATURATED CARBONYL COMPOUND. A 1-ACYLETHENYL ANION EQUIVALENT

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Abstract: Me<sub>2</sub>AlSPh or Me<sub>2</sub>AlSeMe adds to  $\alpha,\beta$ -unsaturated carbonyl compounds in 1,4-fashion. The resulting aluminium enolates react with aldehydes effectively to afford the adducts of the title anion after formal PhSH(or MeSeH)-elimination.

A recent publication<sup>1</sup> on the joining reaction of thiolate anion, an activated olefin and a carbonyl component prompts us to report independent results along the similar line involving (1) 1,4-addition of  $R_2AlX$  to  $\alpha,\beta$ unsaturated carbonyl compound to give the aluminium enolate, (2) aldol reaction<sup>2</sup> of the resulting aluminium enolate, and (3) elimination of HX providing  $\alpha$ -substituted  $\alpha,\beta$ -unsaturated carbonyl compound. The overall transformation provides an access to a 1-acylethenylanion V adding to the aldehyde component<sup>3</sup> to form IV. Among many candidates for X, phenylthio and methylseleno group have been found effective (Table 1).



Run	α,β-Unsaturated Carbonyl Compound	x	Aldehyde RCHO	Method	III Yield <sup>b</sup> (%)	IV Yield from III <sup>b</sup> (%)
1	2-cyclohexenone	SPh	Снасно	A	94	92
2	2-cyclohexenone	SeMec	СНЗСНО	A	77	77
3	2-cyclohexenone	SPh	Сн <sub>3</sub> (Сн <sub>2</sub> ) <sub>7</sub> Сно	A	90	71
4	2-cyclohexenone	SPh	methacrolein	A	97	84
5		SPh	сн <sub>3</sub> сно	A	75	_
6		SPh	сн <sub>3</sub> сно	A	_	50 <sup>e</sup>
7	CH2=CHCOCH3	SPh	Снасно	A	60	57
8	CH2=CHCOCH2	SeMe <sup>C</sup>	снусно	А	55	50
9	2-cyclopentenone	SPh	нсно	А	56	sof
10	2-cyclopentenone	SPh	CH <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> CHO	A	76	60
11	CH <sub>2</sub> =CHCOOC <sub>2</sub> H <sub>5</sub>	SPh	сн <sub>з</sub> сно	Bq	73	87
12		SPh	PhCHO	Bq	77	74

Table 1.	Reaction bet	ween α,β-unsaturated	carbonyl compound	and aldehyde by
	means of Me <sub>2</sub>	Alsph or Me <sub>2</sub> AlseMe <sup>a</sup>		

a. Reactions were performed on 2.0 mmol scale. b. Satisfactory spectroscopic and elemental analyses of all the products. Diastereomer ratios have not been determined. c. Me<sub>2</sub>AlSeMe was prepared <u>in situ</u> according to ref. 4. d. An ate complex Me<sub>3</sub>Al<sup>-</sup> SPhLi<sup>+</sup> was used instead of Me<sub>2</sub>AlSPh. e. Phenylthio group has been eliminated in the reaction mixture without further treatment with CuCl<sub>2</sub> and NaOAc. f. See ref. 7.

To a solution of PhSH (0.26 g, 2.4 mmol) in dichloromethane (2 ml) was added  $Me_3Al$  (1.0 M solution in hexane, 2.4 ml, 2.4 mmol) at 0 °C under argon atmosphere and the reaction mixture was kept there for an additional 20 min. A solution of 2-cyclohexenone (0.19 g, 2.0 mmol) in dichloromethane (2 ml) was added at -78 °C. After the completion of 1,4-addition of thiolate (15 min),<sup>5</sup> the reaction mixture was diluted with tetrahydrofuran (10 ml)<sup>6</sup> and a solution of nonanal (0.34 g, 2.4 mmol) in THF (2 ml) was added dropwise. Stirring was continued at -78 °C for 15 min and the reaction mixture was diluted with ethyl acetate (20 ml) and poured into ice-cooled lN hydrochloric acid. The organic phase was washed with brine, dried, and freed of the solvent. Purification

by column chromatography on silica gel afforded the desired  $\beta$ -hydroxy carbonyl compound III (0.64 g, 90% yield) as a colourless oil (method A).

Unfortunately Me<sub>2</sub>AlSPh was not effective in the reaction of  $\alpha,\beta$ unsaturated esters because of sluggish addition of thiolate and thiol ester formation.<sup>8</sup> The use of a new ate complex<sup>9</sup> Me<sub>3</sub>Al<sup>-</sup>SPhLi<sup>+</sup> has been found to be much more advantageous possibly due to the increased reactivity of thiolate ion itself and the resulting enolate anion corresponding to II.

Spontaneous elimination of phenylthic group in the reaction mixture was observed in run 6. Frequently the conversion of III to IV was achieved by the oxidation (NaIO<sub>4</sub> for SPh and  $H_2O_2$  for SeMe) and successive elimination<sup>10</sup> as given in the Table 1. Moreover, the overall transformation could be performed in one pot, for instance, the addition of cupric chloride (5 mmol) and sodium acetate (5 mmol) before work up in the run 1 afforded the desired compound IV directly in 52% isolated yield.

Treatment of the  $\alpha,\beta$ -unsaturated carbonyl compounds VI<sup>11</sup> and VIII<sup>12</sup> with Me<sub>2</sub>AlSPh provided an efficient method for the cyclization affording VII<sup>13</sup> and IX,<sup>14</sup> respectively.<sup>15,16</sup>



## References and notes

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- 5. Disappearance of the starting enone was monitered by analytical tlc.
- 6. THF was found to be the best solvent for the aldol reaction of aluminium enolate. See also ref. 2.
- 7. This product is a key compound for the synthesis of cyclopentenoid antibiotics (See ref. 3). The research described here was started to exploit an another versatile route to this compound.
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- 9. A hexane solution of Me<sub>3</sub>Al (1.0 mol) was added to a solution of LiSPh derived from PhSH (1.0 mol) and <u>n</u>-BuLi (1.0 mol) in THF at 0 °C. The reagent was used without isolation.
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  The compound VIII was prepared as follows:



- 13. The product VII exhibited: IR(neat) 3480, 1655, 1630(sh) cm<sup>-1</sup>; NMR(CCl<sub>4</sub>)  $\delta$  1.60-2.40 (m, 12H), 2.97 (m, 1H), 4.42 (bs, 1H); exact mass spectrum <u>m/e</u> 166.0980 (calcd for C<sub>10</sub>H<sub>14</sub>O<sub>2</sub> 166.0992).
- 14. A mixture of diastereomers. IR(neat) 3440, 1702 cm<sup>-1</sup>; NMR(CC1<sub>4</sub>)  $\delta$  2.28-2.47 (m, 2H), 3.81 (m, 2H), 7.13-7.48 (m, 5H).
- 15. The reagent Me<sub>2</sub>AlSPh (3.0 mmol) was prepared described above and diluted with THF (9.0 ml). A solution of VI or VIII (1.0 mmol) in THF (1.0 ml) was added over 40 min at 0 °C and the resulting mixture was stirred for an additional 30 min.
- 16. Financial support by the Ministry of Education, Science, and Culture, Japanese Government (Grant-in-Aid #403022), is acknowledged.

(Received in Japan 2 November 1979)